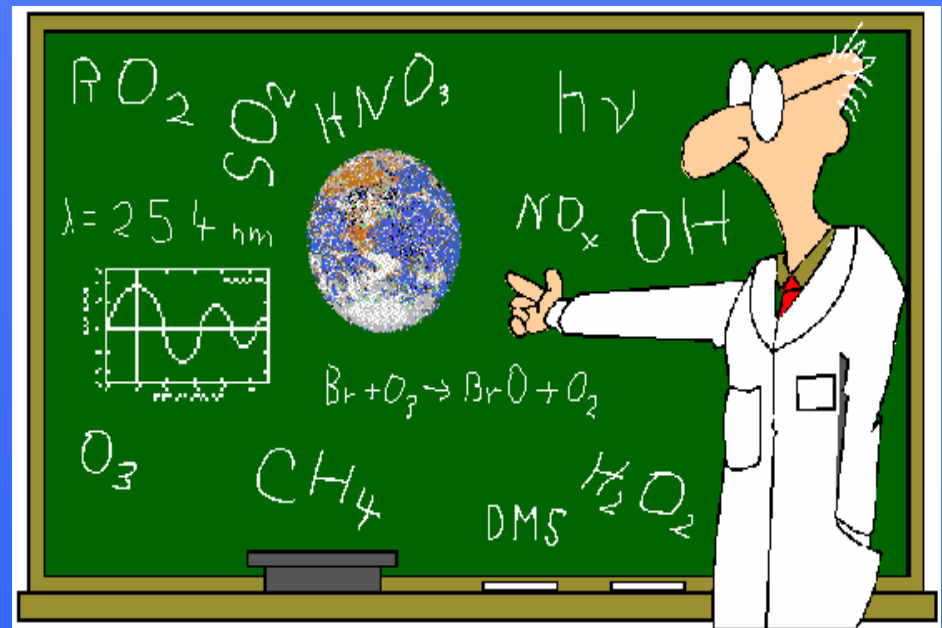


Global cycling of methyl bromide and methyl chloride... and some thoughts on their natural variability

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Dept. of Earth System Science
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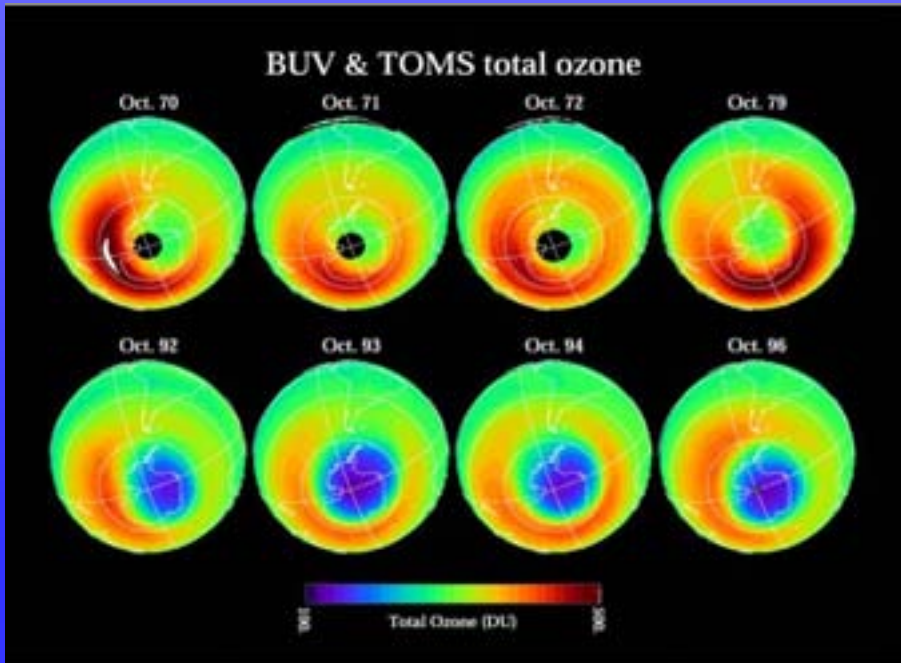
...and many, many others!!!

A quick synopsis for the temporarily impaired....

1. bottom up budgets for methyl halides are very difficult, but they must be done

2. ice cores can provide some answers, and some new questions....but is the past the key to the future?

Stratospheric ozone loss....



Polar Stratospheric Clouds

Type I PSC:	Nitric acid trihydrate (HNO ₃ ·H ₂ O) Ternary solution (H ₂ O, H ₂ SO ₄ , HNO ₃)
Formation Temp:	195 K
Particle diameter:	1 μm
Altitudes:	1024 km
Settling rates:	1 km/30 days
Type II PSC:	Water Ice
Formation Temp:	188 K
Particle diameter:	> 10 μm
Altitudes:	1024 km
Settling rates:	> 1.5 km/day

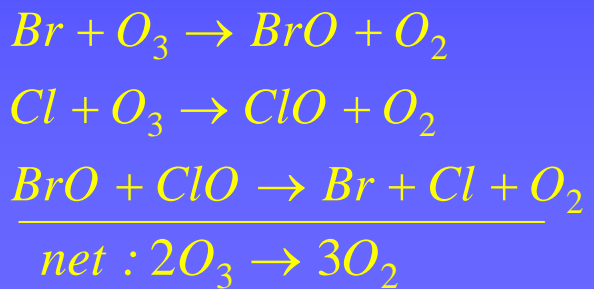
Type II PSC cloud

stratospheric halogen sources...

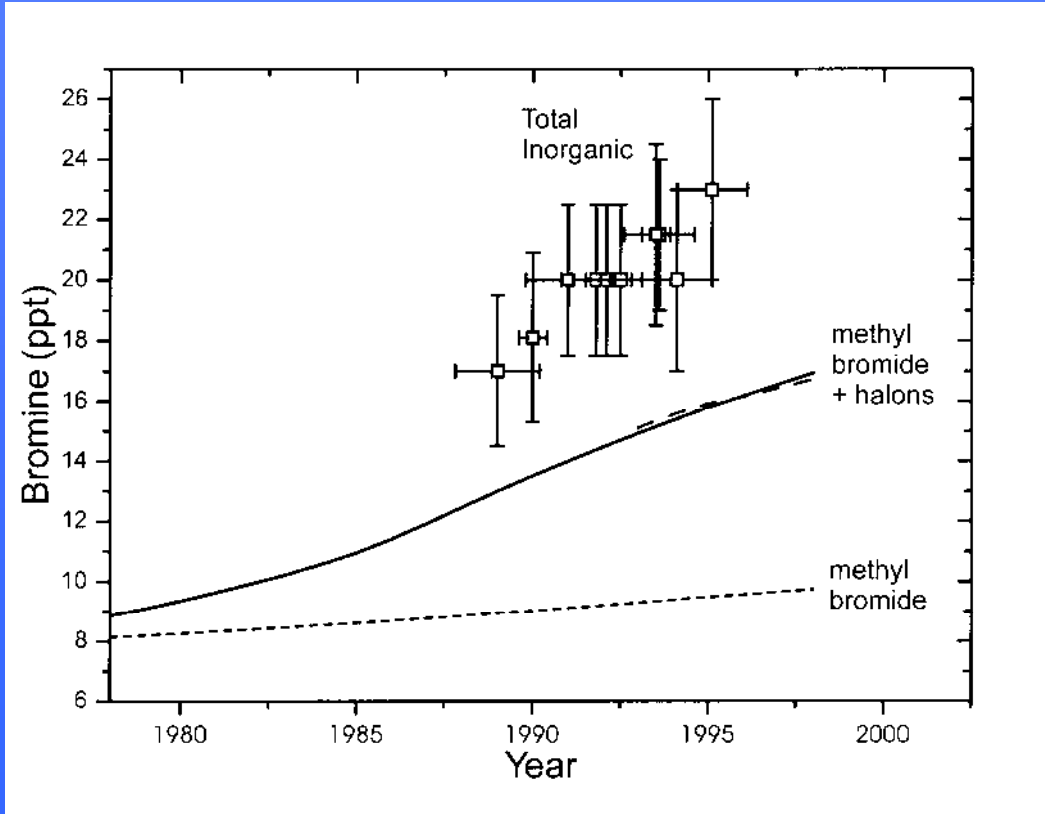
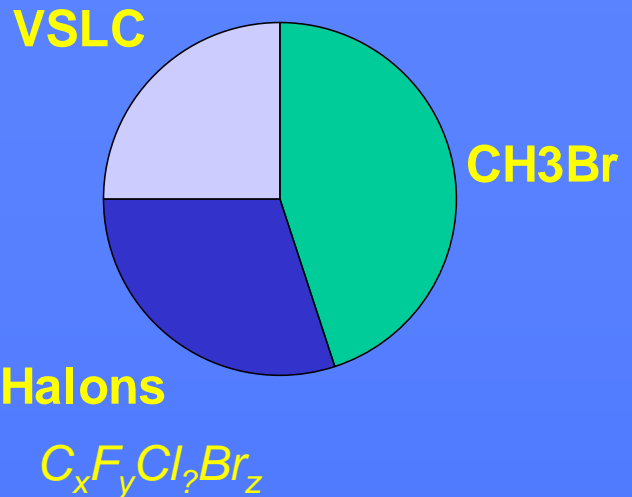
anthropogenic - CFC's, halons,...

and natural - CH₃Cl, CH₃Br, CH₃I, CH₂Br₂, CHBr₃, ...

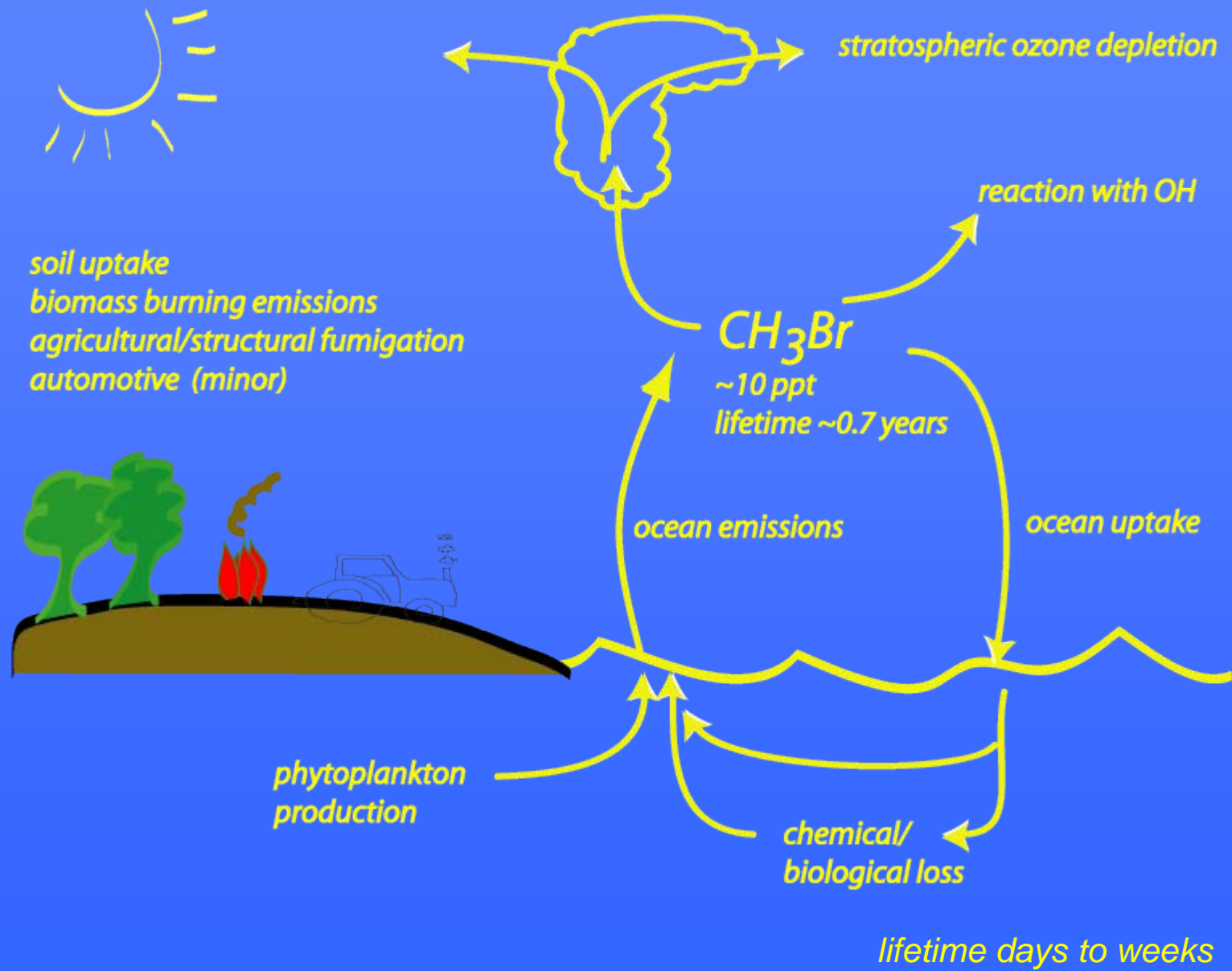
Catalytic ozone destruction
Br is ~50x more effective than Cl!



Methyl bromide and stratospheric bromine ...



Global CH₃Br cycle



Question #1:

How will the atmosphere respond to phase-out of agricultural use of CH_3Br under the Montreal Protocol?

need to know:

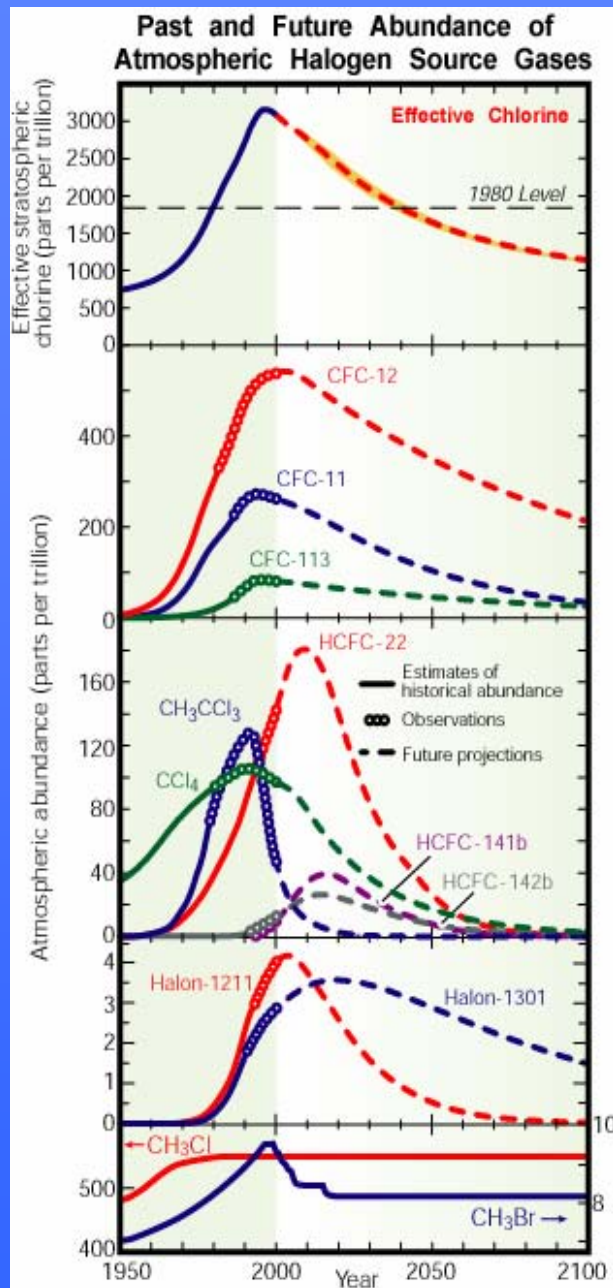
- *the total budget (i.e. all sources/sinks)* ✓
- *fraction of anthropogenic sources* ✗
- *response of the ocean (“buffering capacity”)* ~✓
- *atmospheric lifetime (0.7 years via OH, soils)* ✓

Question #2: Global climate change

How will atmospheric CH₃Br respond to future changes in global climate?

- *changing climatic conditions means changes in both physical and biological systems*
- *predicting the response of atmospheric methyl halides even to simple physical forcing requires understanding the dynamics of the system, not just the current net fluxes.*

need to know: ...basically everything!



CH₃Br saturation anomaly vs SST

$$\Delta_g = \text{Saturation Anomaly} = \frac{(C_W / H - p_a)}{p_a} \times 100$$

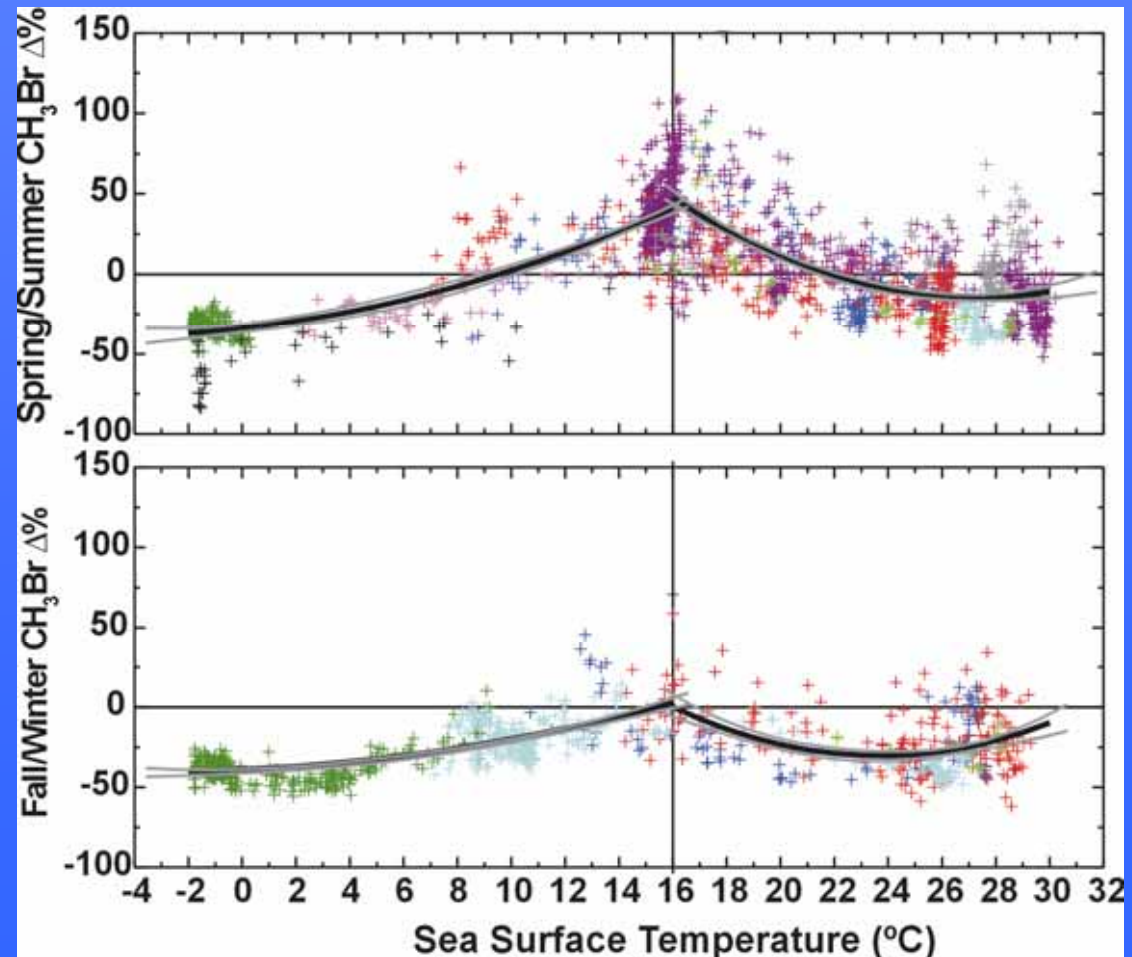
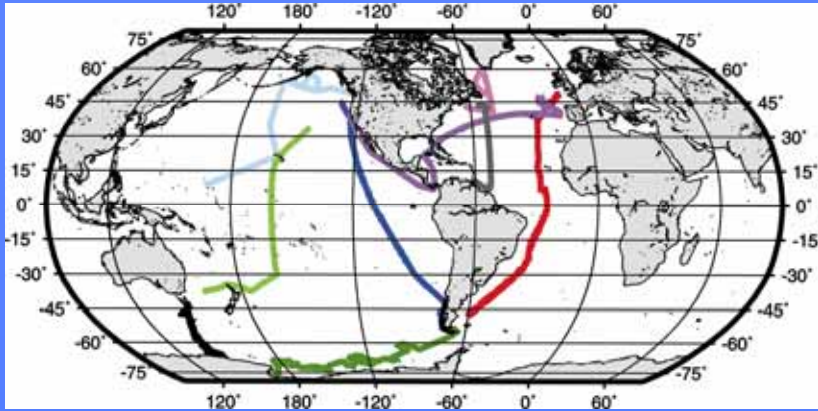
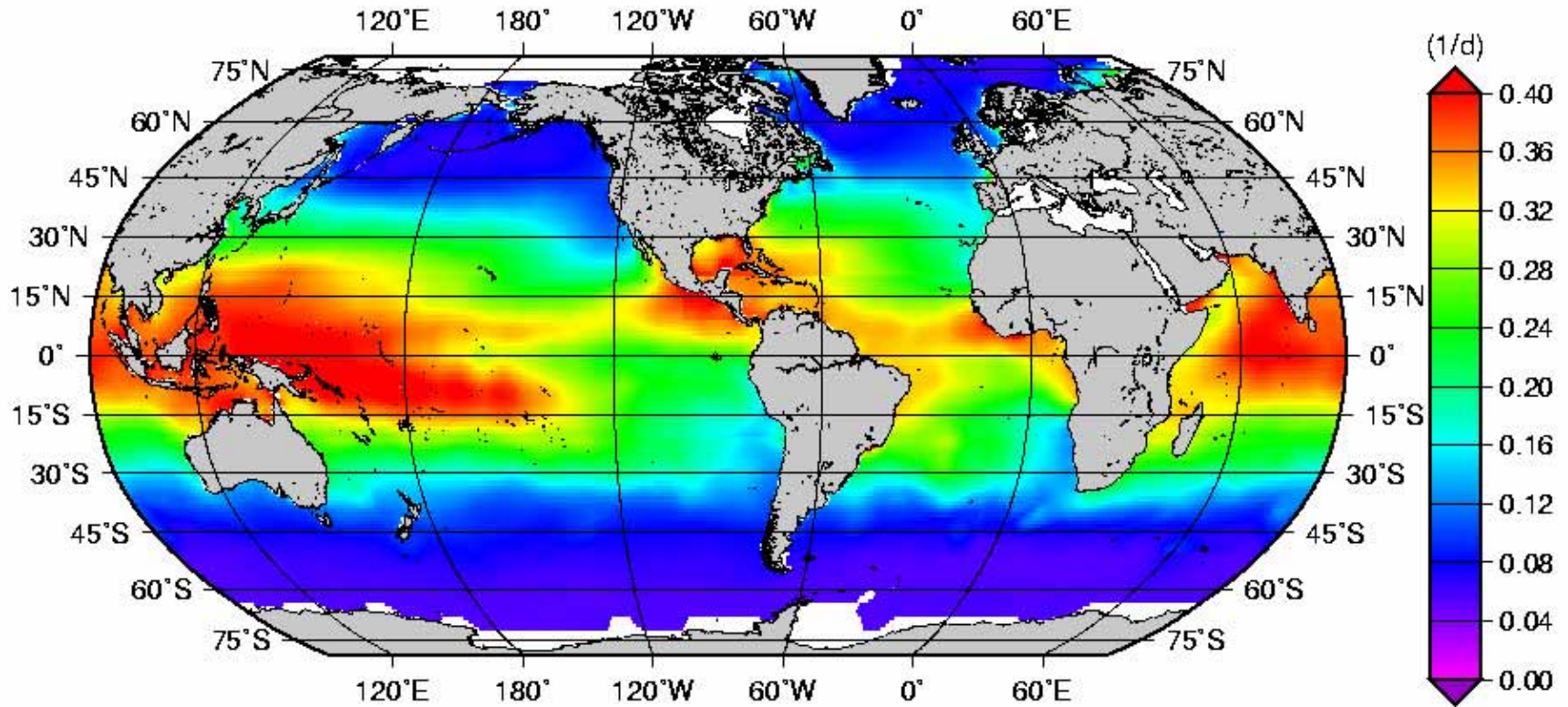


Figure 1. Nine cruises of CH₃Br Δ_g data with a map showing cruise tracks where Δ_g data was collected: BLAST I, BLAST II, BLAST III, GasEx98, RB-99-06, CLIVAR SR3, GM98A, GM98P, and G99 (a), CH₃Br saturation Δ_g data from the nine cruises plotted against SST and separated by season with spring/summer (b) and fall winter (c). Quadratic fits (—) are used to describe the data. Figure and data from King *et al.* [2002] and references therein.

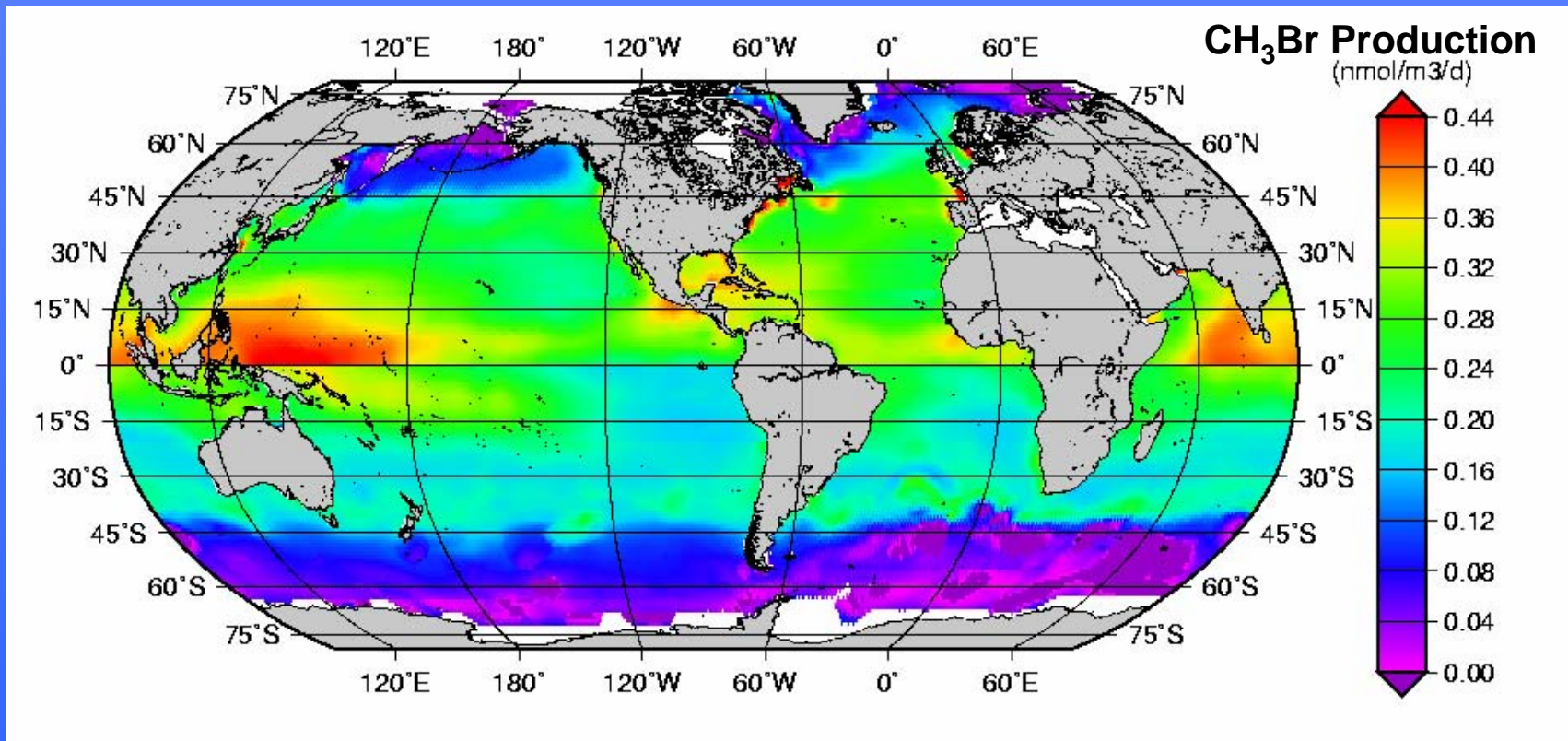
CH_3Br loss rate constant



chloride substitution/hydrolysis
biological (bacterial?) uptake

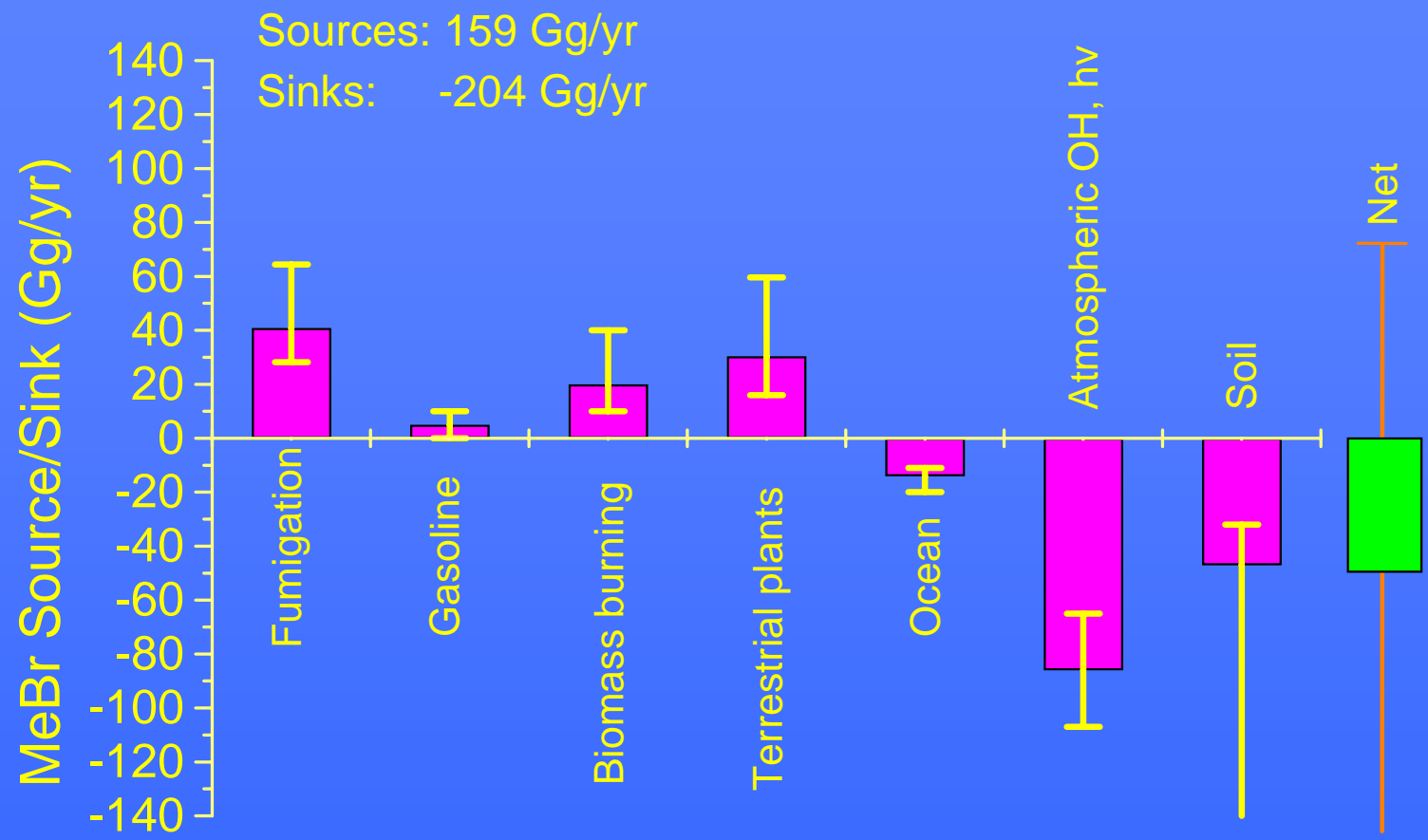
Inferred distribution of annually averaged CH_3Br oceanic production rates ($\text{nmol m}^{-3} \text{d}^{-1}$)

$$\text{Production} = \frac{K_W}{z} \left(\frac{\Delta_g - \Delta_{f11}}{100} \right) \frac{p_a}{H} + (k_{chem} + k_{biol}) C_W - \frac{D_z}{z} \frac{\Delta C}{\Delta z}$$



organisms responsible?
biochemical pathways?

Atmospheric CH₃Br budget

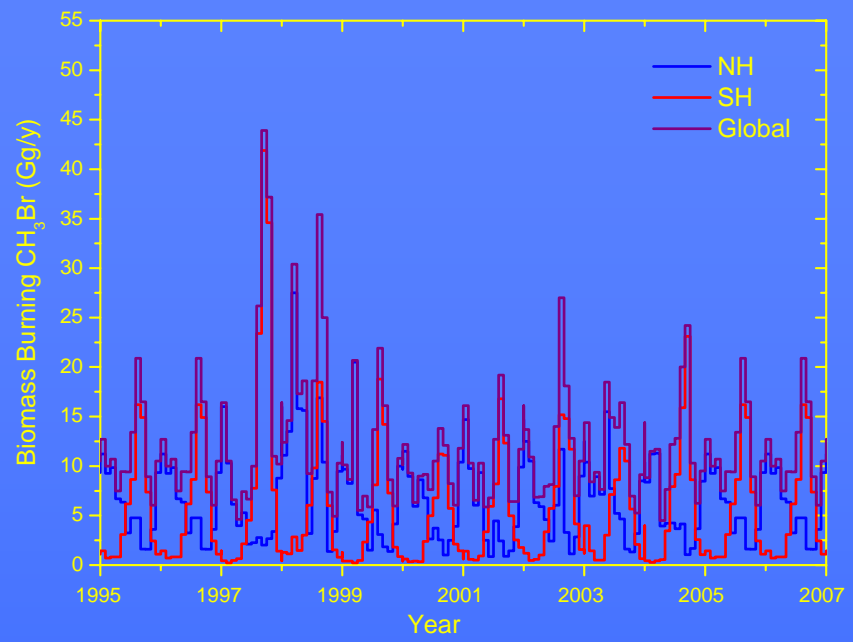
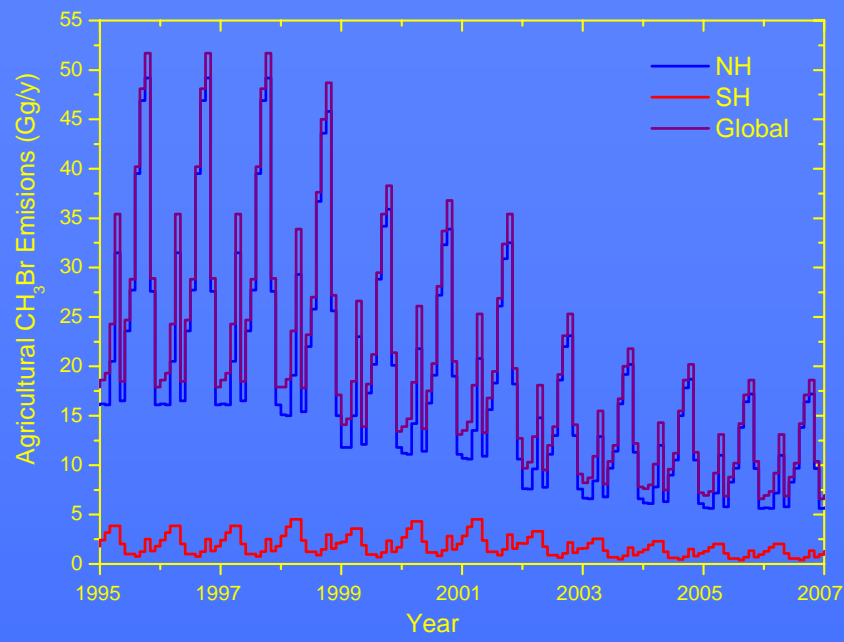


Note: The oceanic contribution to atmospheric CH₃Br is the net effect of a large oceanic source (56 Gg/yr) and a large oceanic sink (77 Gg/yr) !

Source: WMO 2002

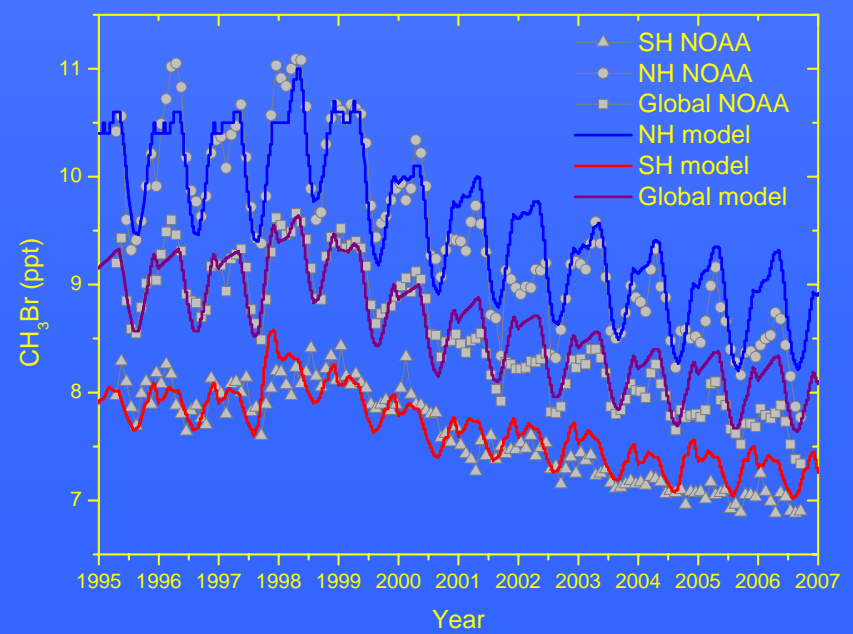
So what?

CH₃Br – recent trends (1995-2006)

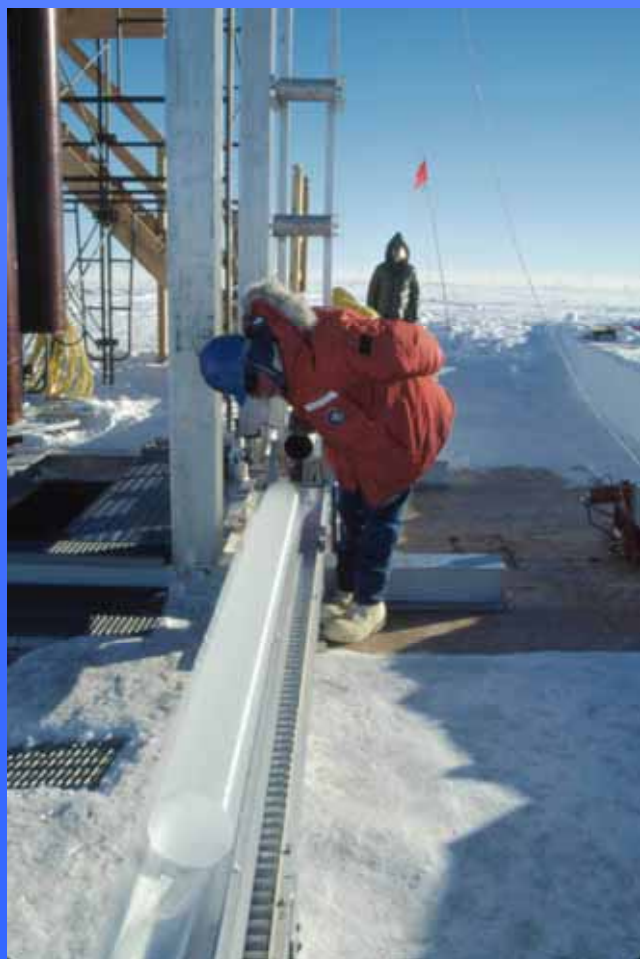


- Agricultural emissions
- Biomass burning

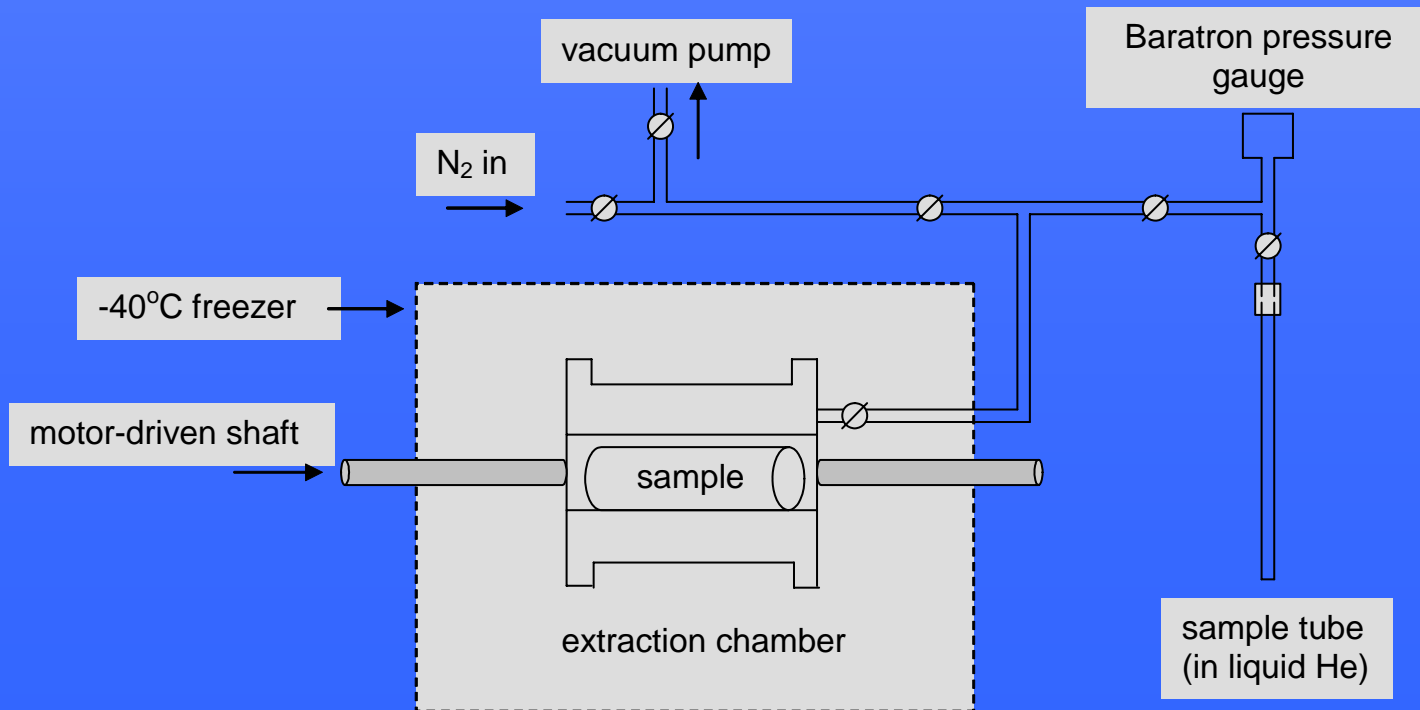
Model vs observations... →



... to the ice core record!



Dry extraction of trace gases from ice cores...



Sample Request # 2002-028

Cut for: Dr. Eric Saltzman

Core: SDMA

Section: 74

Tube: 74

TD: 65.416

BD: 65.616

Cut: MCA.02

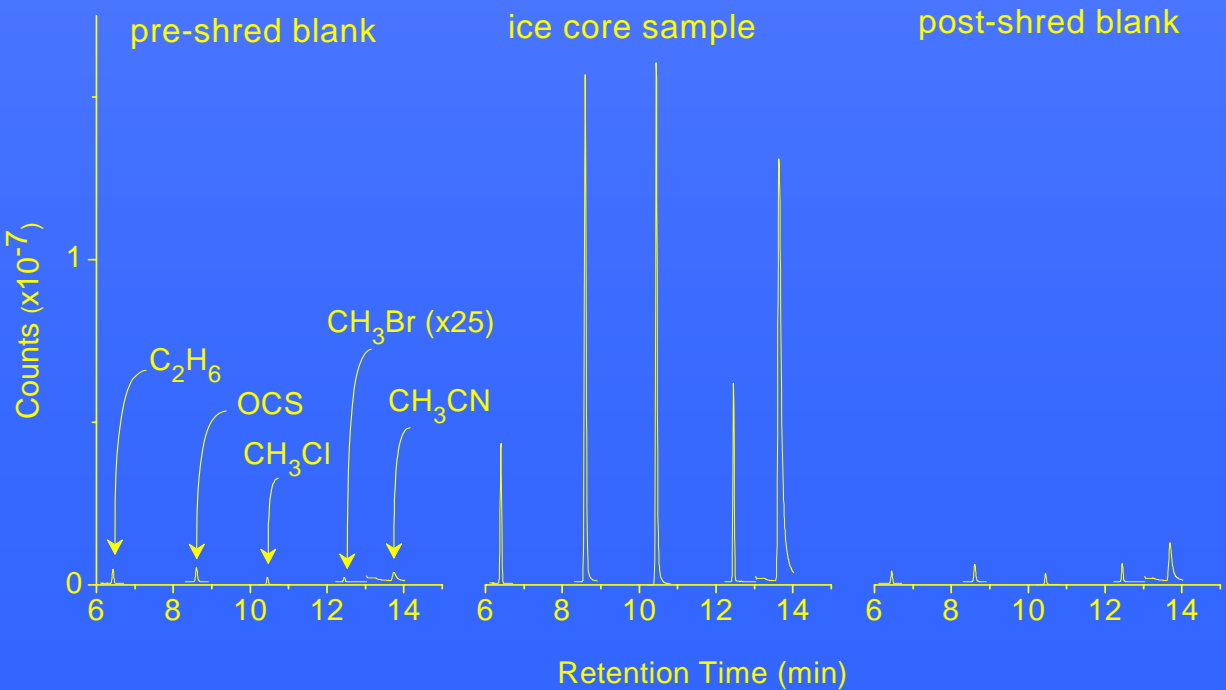
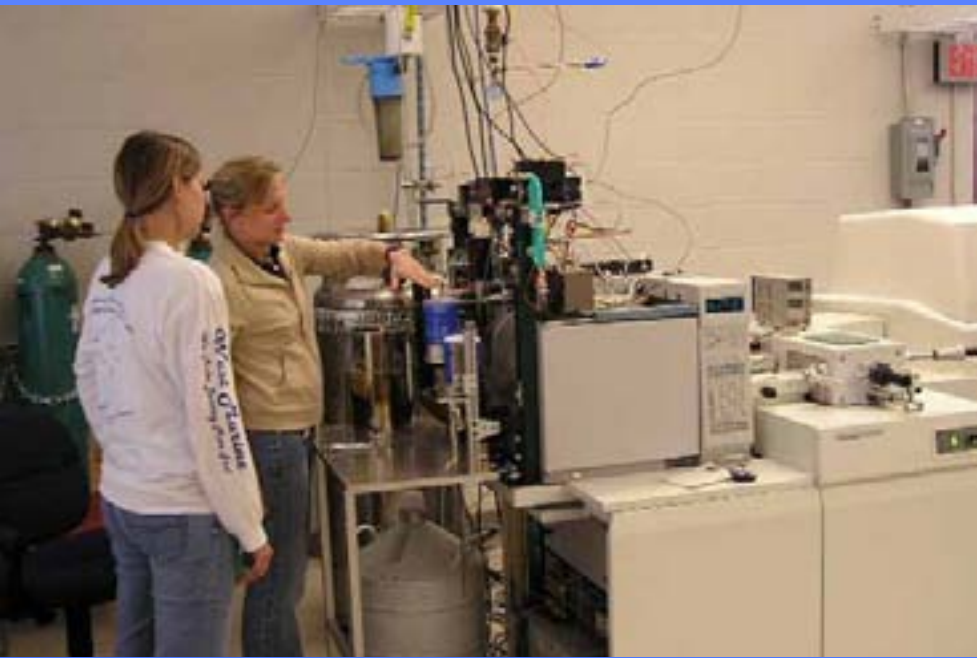
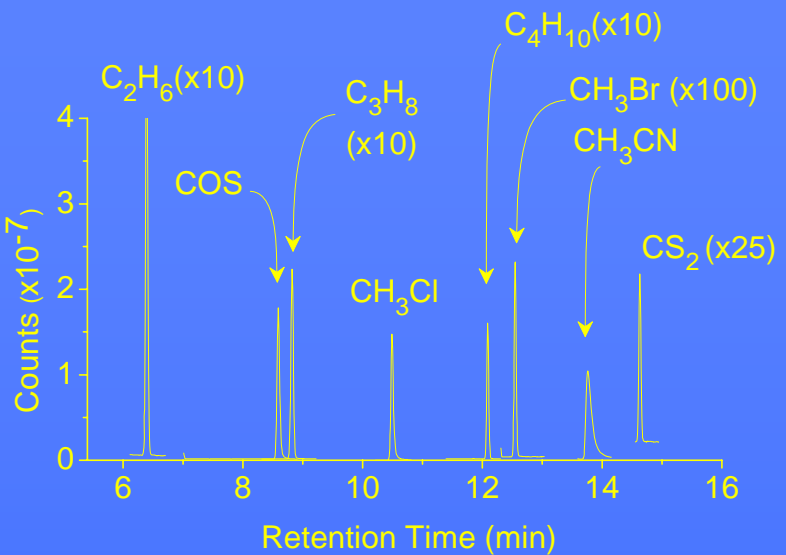


Length: .2m.

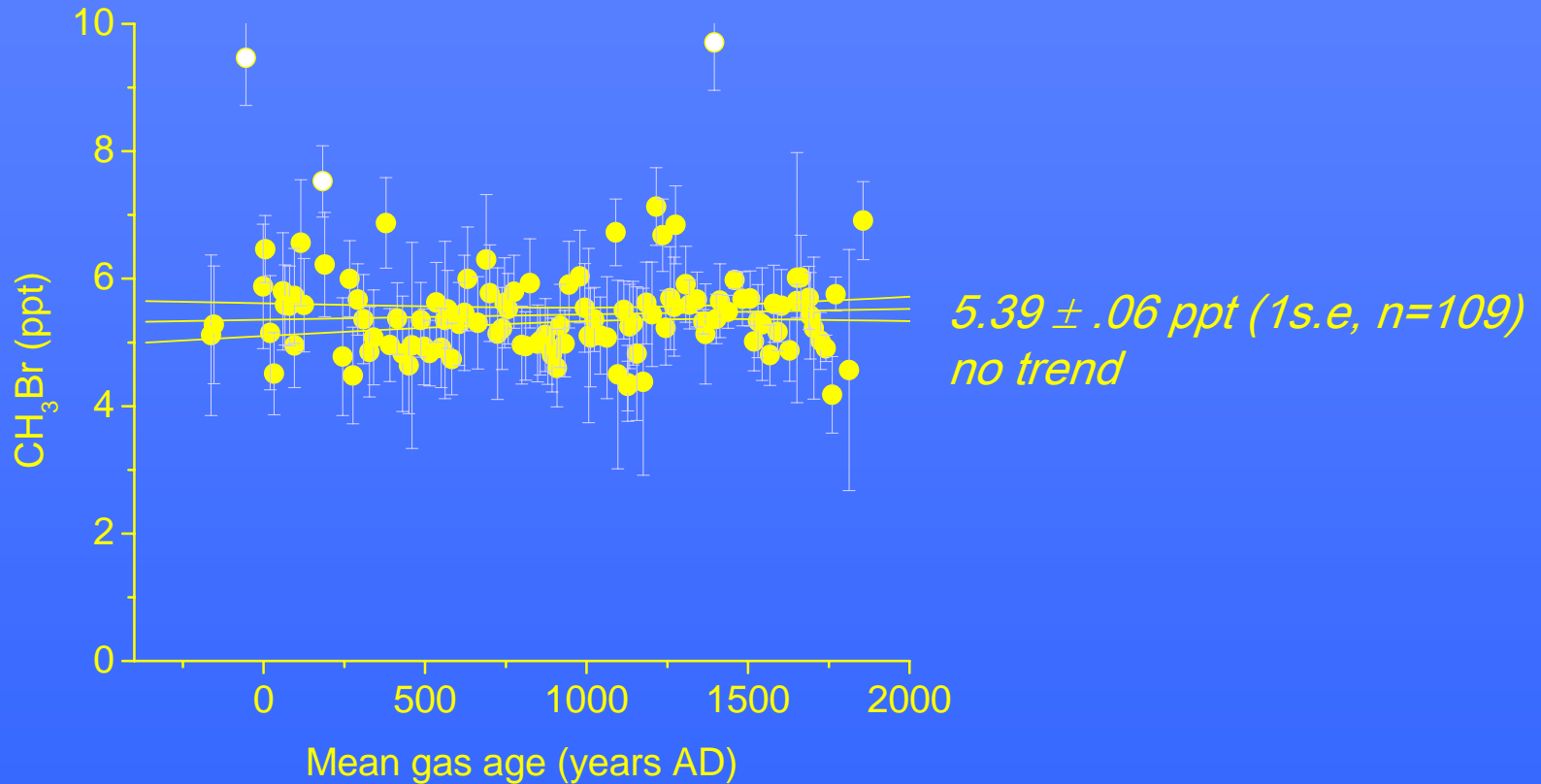
Cut by: Todd Wojcik

1/27/2003

Ice core trace gas analysis...

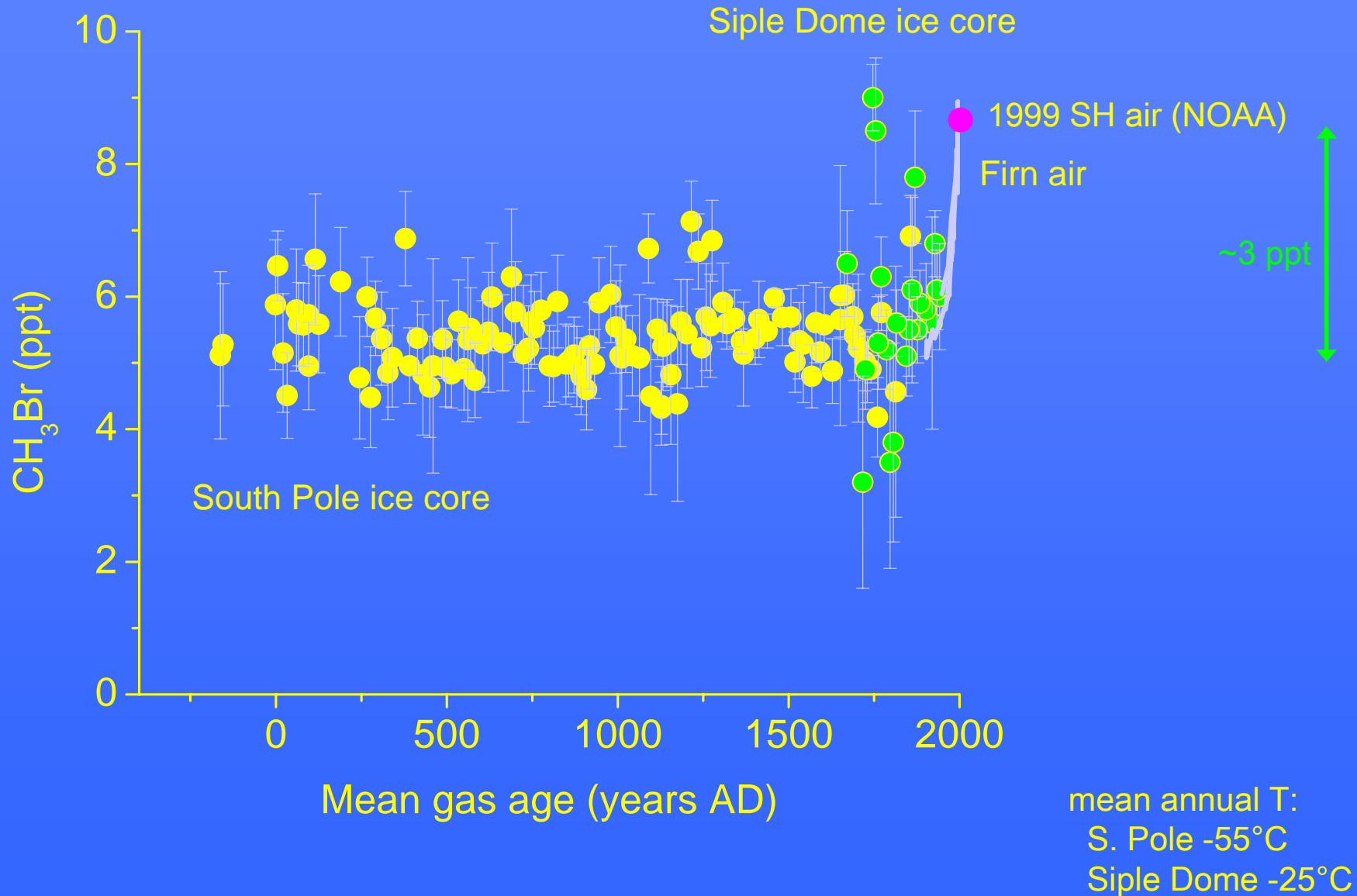


2,000 year record of atmospheric CH_3Br from a South Pole ice core

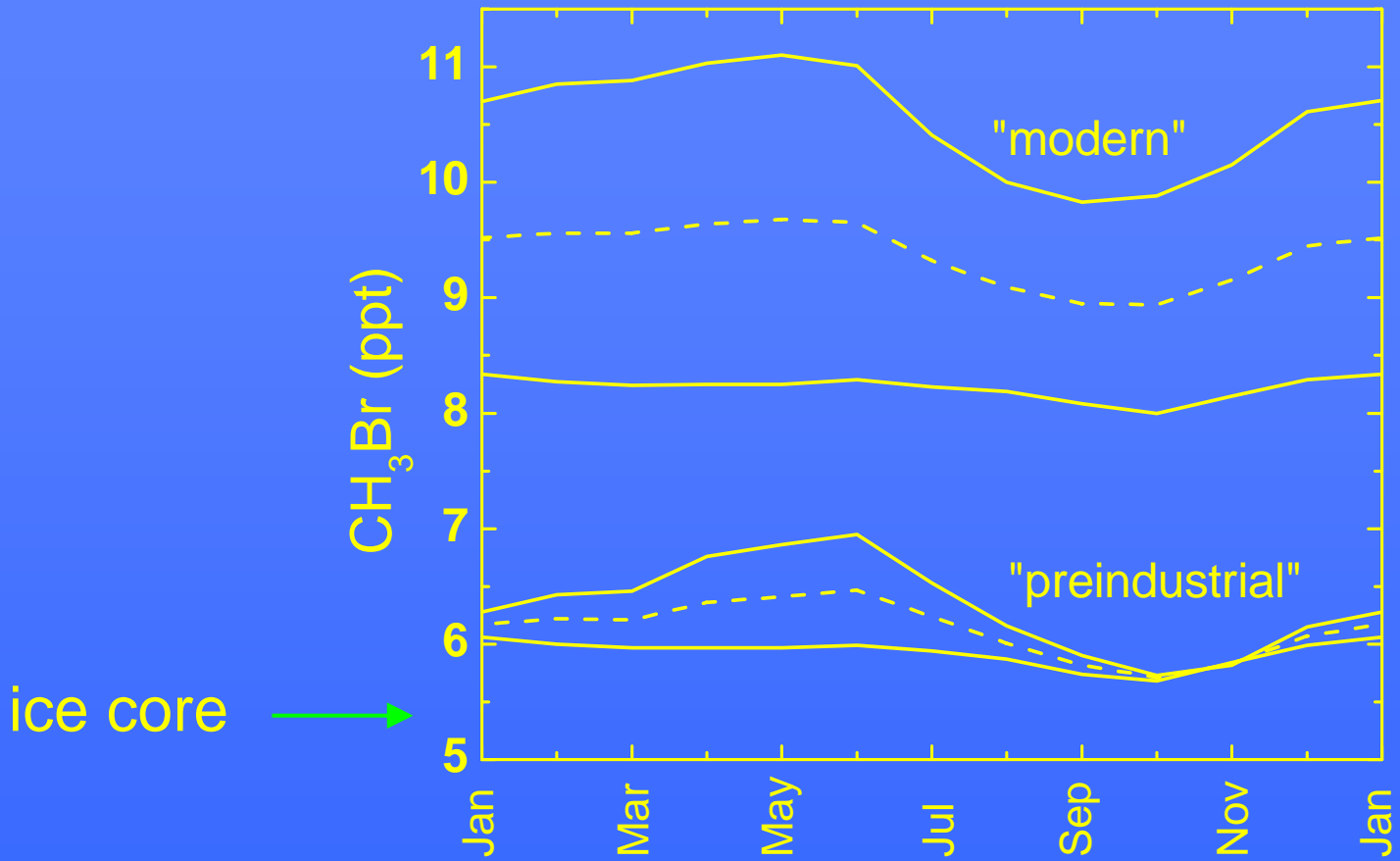


(note to paleoceanographers: time marches to the right...)

Man's impact on the CH_3Br cycle...

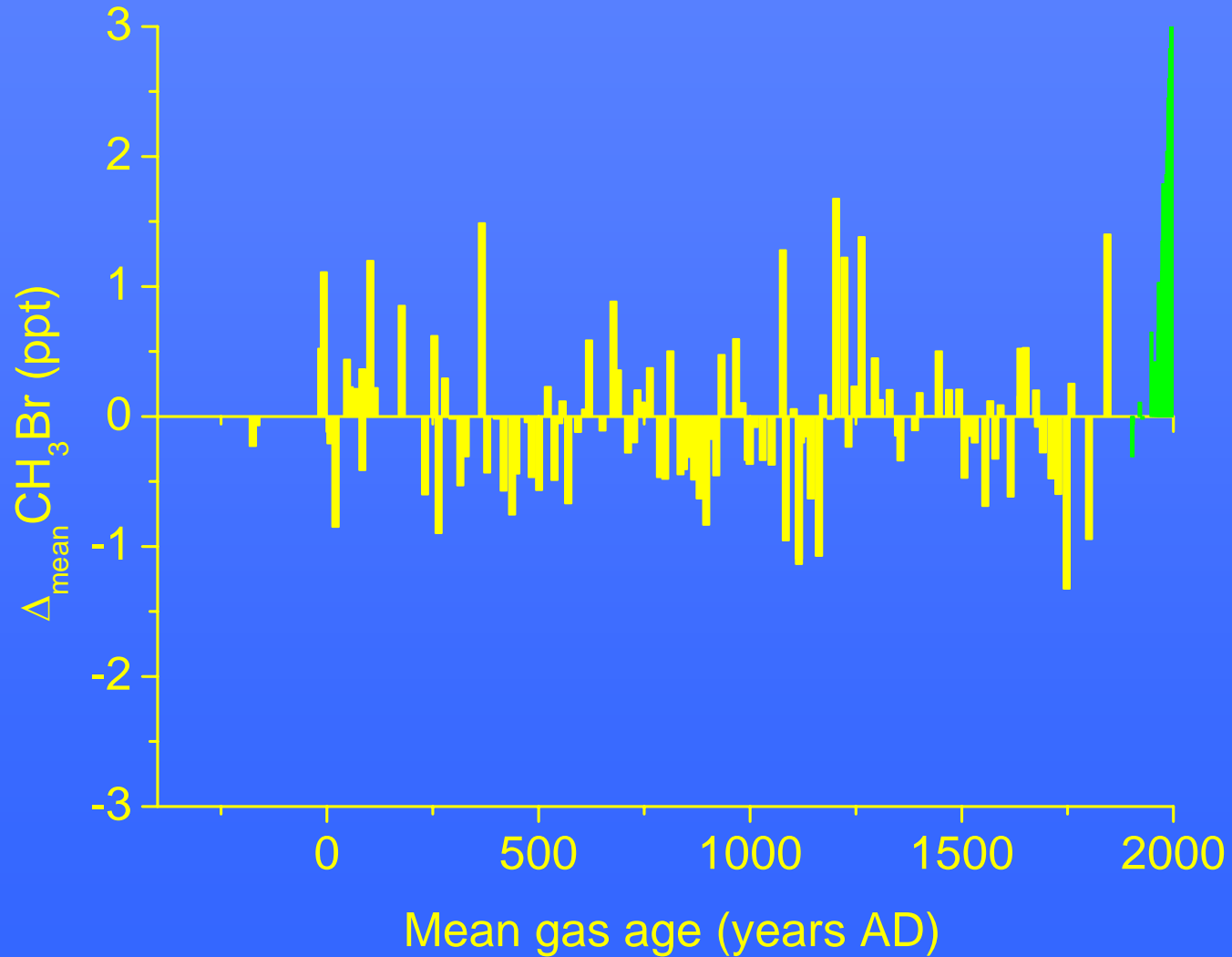


Model simulations...



reasonable, assuming SH missing source is natural and nothing else changed...

Evidence of centennial-scale variability... or noise?



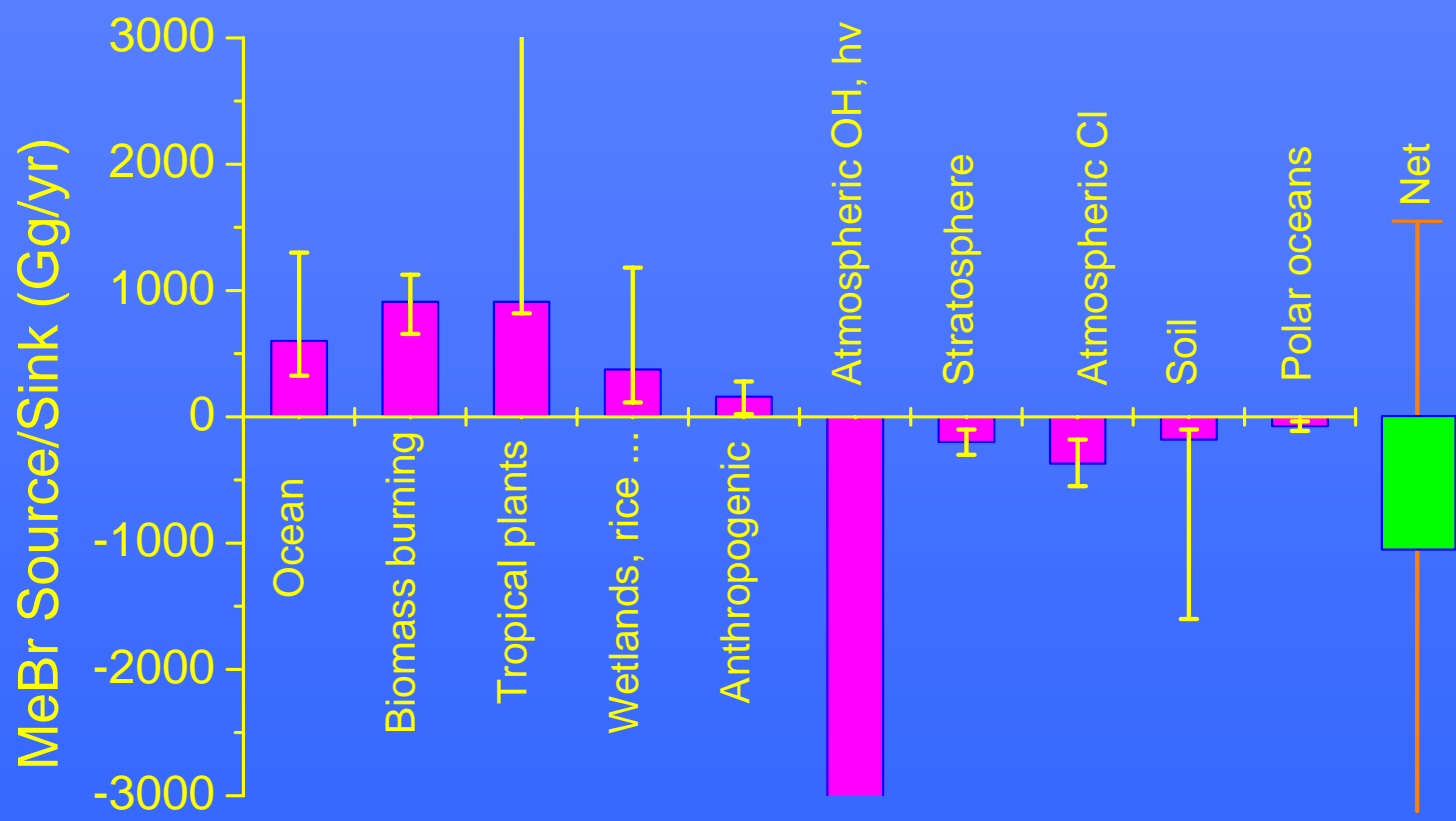
Some points to consider:

- *CH₃Br seems to be stable in polar ice!*
- *Southern hemisphere CH₃Br levels ~30% lower for past 2ky prior to industrialization*
- *Natural variability might be significant, but ...*
 - *it's awfully close to experimental uncertainty*
 - *Late Holocene climate change was modest*
- *The phase-out of CH₃Br will probably result in a 3 ppt SH decrease (4.5 ppt global; 225 ppt Cl equivalent).
That's a good thing.*

CH₃Cl global budget

mixing ratio 550 ppt
atm lifetime ~1 year

Sources: 2,956 Gg/yr
Sinks: -1,049 Gg/yr

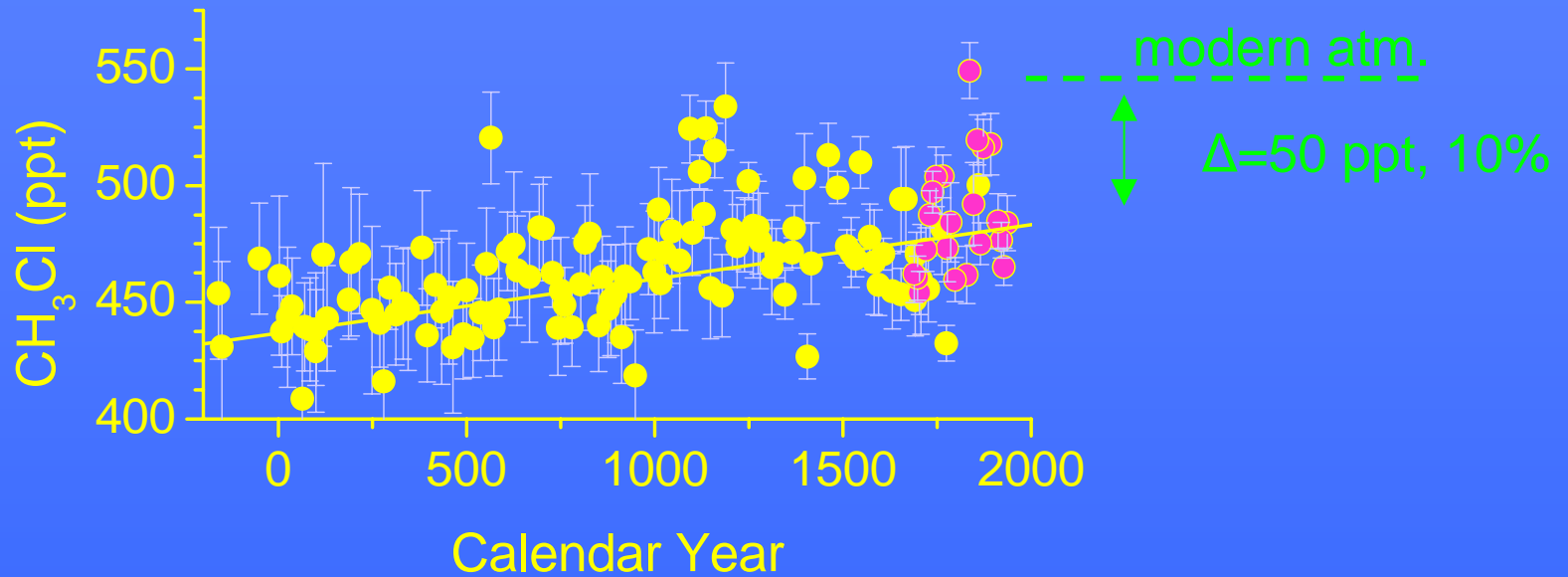


Source: WMO 2002

ice core methyl chloride...

South Pole ice core ●

Siple Dome ice core ●



trend: 3 ppt/100 years

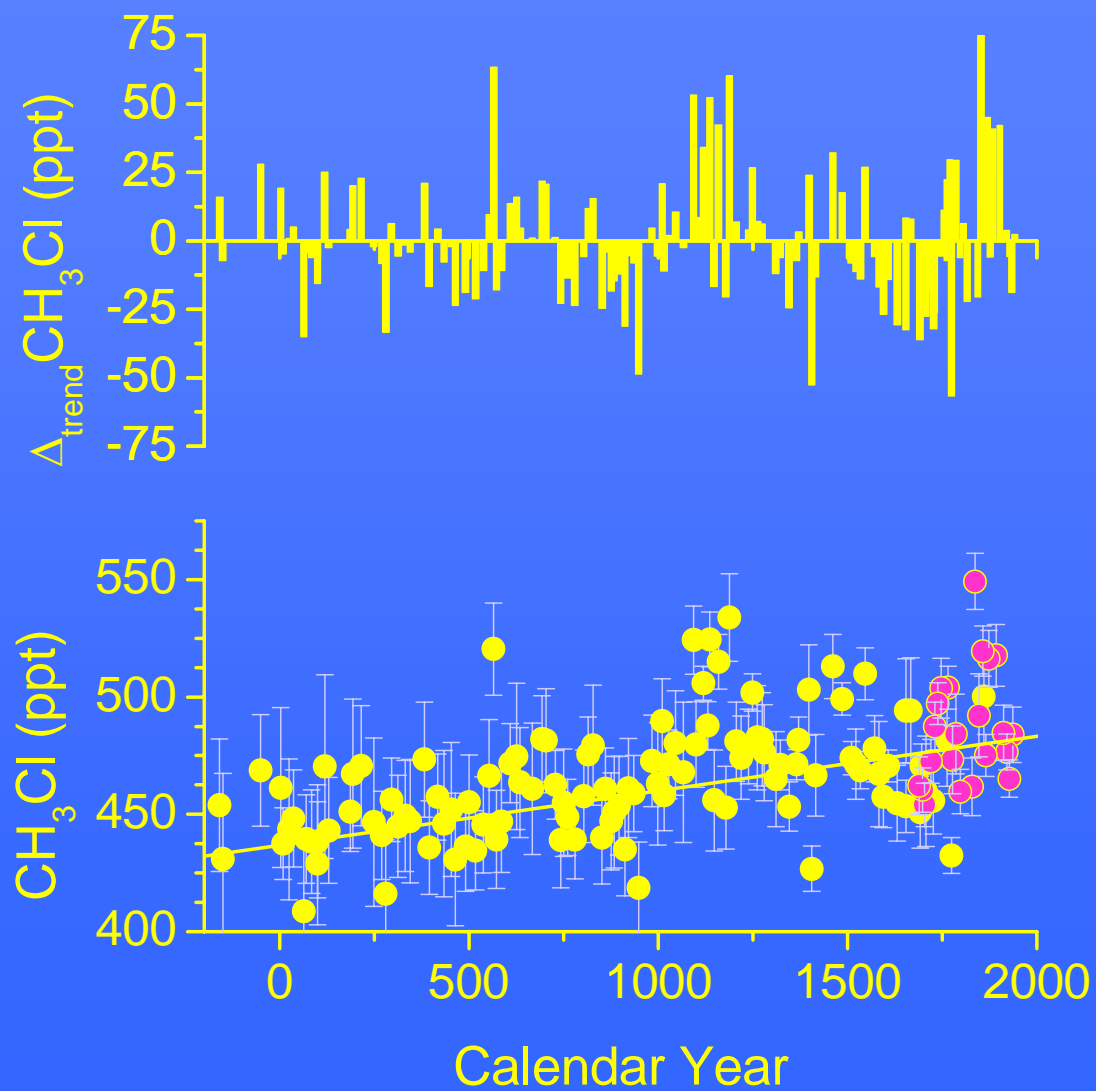
Long-term variability or downcore loss?

Williams et al., in press

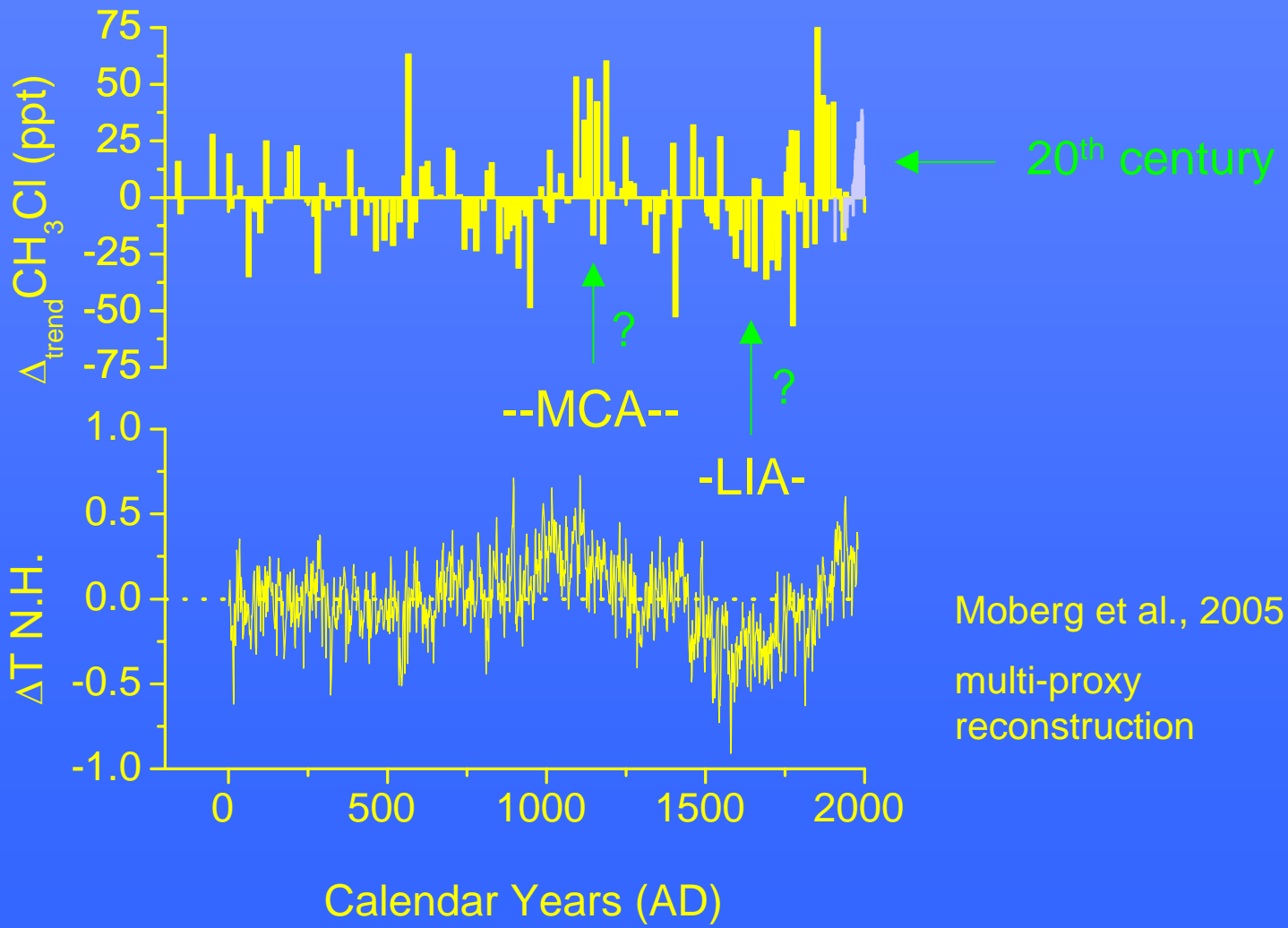
ice core methyl chloride...

South Pole ice core ●

Siple Dome ice core ●



ice core methyl chloride... climate-controlled variability?



Take home messages...

- CH_3Cl appears climate-sensitive, and CH_3Br might be. Is this an ocean signal, a terrestrial signal, or both?
- Global warming may increase CH_3Cl and CH_3Br levels by >100 ppt Cl_{eq} . *That's probably not a good thing.*
- caveat...in the future, tropical land use changes might trump climate-signals
- In a warmer world, will atmospheric halogens (and stratospheric ozone) return to their preindustrial state? *Probably not.*